

Simultaneous Estimation of Ciprofloxacin and Tinidazole by U.V Spectrophotometer using a Hydrotropic Solubilization Technique

Km. Babita^{1*}, Kapil Kalra¹, Neetu Rajpoot¹,
Poonam Joshi² and Vandana Pokhriyal¹

¹Alpine college of management and technology, Dehradun Uttarakhand India.

²Uttaranchal Institute of Pharmaceutical Sciences, Uttaranchal University,
Dehradun Uttarakhand India.

*Corresponding Author E-mail: babitanegi495@gmail.com

<https://dx.doi.org/10.13005/bpj/2621>

(Received: 13 May 2022; accepted: 14 November 2022)

Advanced, easy and accurate UV spectroscopy technique (I & II) was developed for estimation of Ciprofloxacin and Tinidazole in bulk and tablet dosage form and validated using hydrotropic solubilization techniques. In Method I we focused on solving the simultaneous equation on the basis of absorbance measurement at 270nm and 320nm wavelength of Ciprofloxacin and Tinidazole individually in 2M urea solution which results in enhancement of solubility of CPH and TNZ to about 15 folds as compared to distilled water. Strategy II is lay out idea of Q-analysis in which absorbance was estimated at 288nm (iso-absorptive point) 270nm (?max of CPH) in 2M urea. Ciprofloxacin and Tinidazole display linearity at those frequencies and comply with Beer's regulation in the focus scope of 0.2-1.2µg/mL and 10 to 60µg/mL separately. Recovery reads up for CPS and TNZ were completed and rate recuperation of the medications gotten in the scope of 99.2 - 99.6% (Method A) and 98.0-100.8% (Method B) affirming the exactness of the proposed strategy. Both the techniques showed great reproducibility and recuperation with %RSD less than 2. Factual approval of the information shows that the proposed strategies can be effectively applied for the standard examination of medications in business tablets.

Keywords: absorbance ratio method; hydrotropic agent; Simultaneous estimation method.

Hydrotropic is solubilization process, where the adding of huge quantity of a subsequent solute outcome in the rise of aqueous solubility of insoluble and slightly solvable drugs. A poorly water- soluble drug refers to a “practically insoluble”¹. Poor solubility of drugs is a crucial factor in dosage form designing and due to which drugs show low bioavailability. Hydrotropic agent like, nicotinamide, lysine, β-cyclodextrin and urea, can be utilized for the solvency upgrade². synthetically ciprofloxacin be in to the gathering of manufactured fluoroquinolone anti-infection agents

with strong action against wide range microscopic organisms. Tinidazole is dynamic against both gram+ and gram - microbes.

MATERIAL AND METHODS

Chemical and Reagent

Analytical grade of reagents and solvents take place used for the quantitative estimation of Ciprofloxacin and Tinidazole from Himgiri traders, whereas gift sample of Ciprofloxacin

and Tinidazole was obtained from east African overseas. 2.0M urea was used as hydrotropic agent.

Instrumentation

Double beam UV-Visible Spectrophotometer of Agilent Technologies (Model No- G6860A) take placed. All weighing done on electronic balance.

Selection of solvent

Solubility of the two not set in stone at $28 \pm 2^\circ\text{C}$. An immense measure of medications was added to the receptacle containing different hydrotropic arrangement. The measuring glasses were shaken precisely for 12 hrs. at $28^\circ\text{C} \pm 10^\circ\text{C}$ in a mechanical stirrer. Sample were permitted to get stable within 24 hrs. and afterward agitate for

5min at 2000 rpm. The supernatant fluid taken for proper dilution after separated through filter paper (whatmann) and investigated spectrophotometric partner against reference. After examination it was observed that the expansion in the dissolvability of CPH and TNZ was viewed as in excess of 15 folds in 2.0M urea arrangement when contrasted with solvency in refined water^{5,6}.

Preparation of Stock Solution (1000µg/ml)

Accurately weigh 50mg CPH and TNZ independently and moved to two separate 50ml volumetric, dissolve with the 2M urea arrangement and last volume left up to mark with distilled water get stock soln. of CPH (1000µg/ml) and TNZ (1000µg/ml).

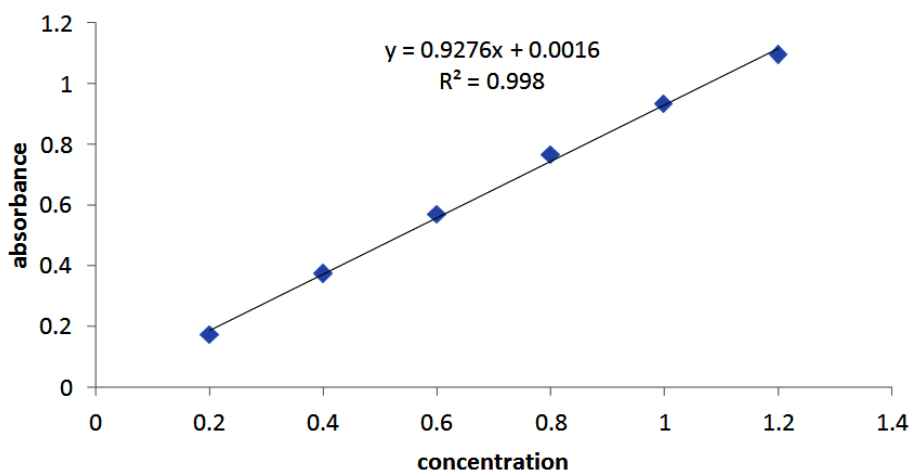


Fig. 1. Calibration curve of ciprofloxacin at 270nm

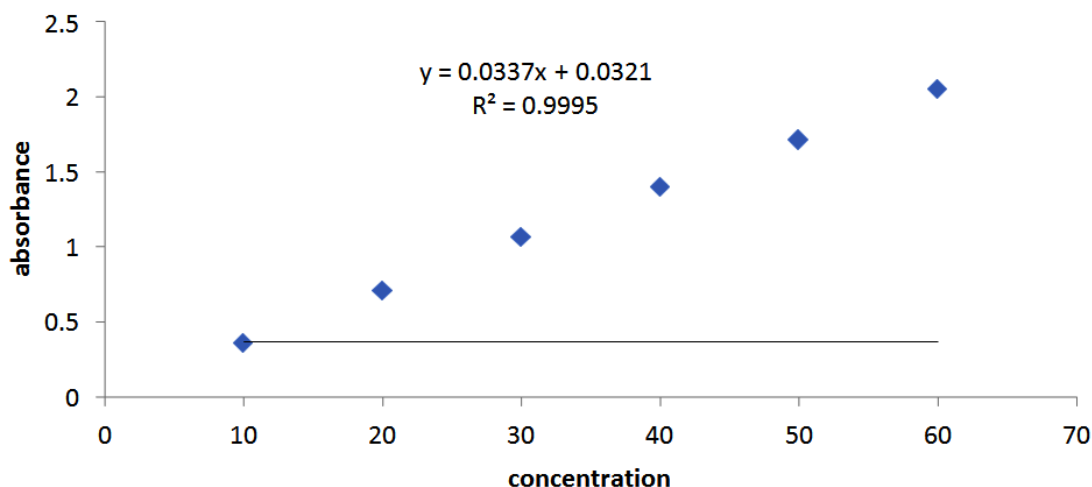


Fig. 2. Calibration curve of Tinidazole at 320nm

Preparation of Working Standard (100µg/ml)

From the above stock solution, 5ml every one of CPH and TNZ was taken more time, to isolate 50ml volumetric carafe and volume was up to 50ml D.W water to acquire working norm of CPH(100µg/ml) and TNZ(100µg/ml).

Method I: Simultaneous Equation Method

Assuming an example contain two absorbing drug (X and Y) all absorbed at different ϵ_{max} . it very well may be practicable to choose the two drugs by the simultaneous equation^{7,8}.

Determination of wavelength of maximum absorbance of Ciprofloxacin and Tinidazole

From the working standard, 10µg/ml solⁿ of CPH and TNZ independently prepared in D.W and the solutions were scanned under spectrum mode for 200-400nm wavelength range against blank to analyses the ϵ_{max} . Sharp peaks were detected at 270nm for CPH and 320nm for TNZ.

Preparation of Calibration curve of CPH and TNZ

From the working norms of CPH(50µg/ml) 0.2, 0.4, 0.6, 0.8, 1.0, 1.2 and TNZ(50µg/ml) 10, 20, 30, 40, 50, 60 ml were moved to 10ml volume made up in volumetric flask to mark with D.W. Calibration curve of absorbance versus concentration was plotted and the regression coefficient was determined as shown in fig 1 and 2.

Both the drugs obey's Beer-Lambert's law in the conc. of 0.2- 1.2µg/ml for Ciprofloxacin and 10-60µg/ml for Tinidazole. The correlation coefficient were found to be $R^2 > 0.999$. The absorptivity values of CPH and TNZ are calculated at their respective wavelengths and presented in table 1.

Derivation of equation ^[9]: from the absorptivity values obtained for CPH and TNZ,

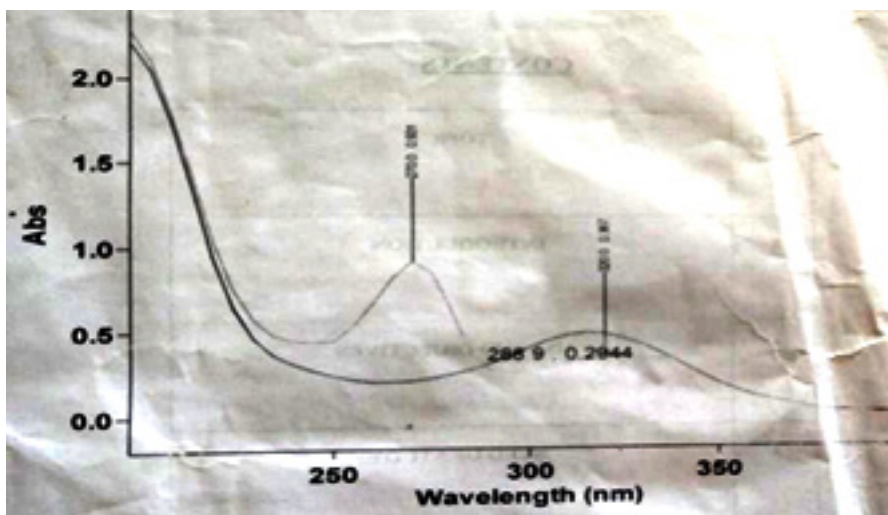


Fig. 3. Overlay spectra of ciprofloxacin and tinidazole

Table 1. Absorptivity values (A 1 %, 1 cm) of CPH and TNZ by method A and B

Conc. µg/ml	Method I		Method II								
	CPH 270nm	CPH 320nm	Conc. µg/ml	TNZ 270nm	TNZ 320nm	CPH 270nm	CPH 288nm	Conc. g/ml	TNZ 270nm	TNZ 288nm	
0.2	865	471	10	354	347	0.2	771	180	10	431	161
0.4	680	397	20	226	315	0.4	646	197	20	256	132
0.6	646	433	30	188	262	0.6	590	170	30	223	108
0.8	614	467	40	163	207	0.8	578	133	40	195	146
1.0	598	454	50	159	188	1.0	564	100	50	168	194
1.2	578	469	60	149	204	1.2	562	167	60	150	183
mean	662	445		206	253	mean	618	157	mean	237	154

simultaneous equation is derived for determination of these two drugs in combination in their pharmaceutical formulations.

$$A1 = 662CCPH + 206CTNZ \quad A2 = 445CCPH + 253CTNZ$$

A1 and A2 are the absorbance of mixture at chosen wavelength 270nm and 320nm¹⁰.

Method II: Absorbtion Ratio Method

Selection of Iso-absorptive point

From the overlay spectrum of CPH and TNZ two wavelength were selected, one at 288nm (λ1) isoabsorptive point for both drugs and other at 270nm(λ2) wavelength of ciprofloxacin for the study. Overlay spectrum is shown in figure 2

and the absorptivity values for CPH and TNZ at selected wavelength are presented in table 1.

Derivation of equation¹³: In Q analysis method, from overlay spectra of CPH and TNZ absorbances were detected at chosen wavelength i.e. at 270nm (λmax of CPH) and 288nm (isoabsorptive point). The concentration of each drug in the tablet is determined by substituting the absorbances and absorptivity in the equation

$$CCPH = (1.97-1.33) / (4.21-1.337) \times A1 / ax1$$

$$CTNZ = (1.97-4.21) / (1.337-4.21) \times A2 / ay1$$

A1 and A2 absorbance of samples at 270nm and 288nm. ax1 and ay1 are the absorptivity of CPH at 270nm and TNZ at 288nm.

Table 2. Determination of accuracy of CPH and TNZ

Drug	Level of recovery (%)	Quantity taken (µg/ml)	Quantity added (µg/ml)	Total (µg/ml)	Quantityrecovered		% Recovery(n=3)	
					Method I	Metho d II	Method I	Method II
CPH	80	5	4	9	9.1	9.2	101.1	102.2
	100	5	5	10	9.8	9.6	98	96
	120	5	6	11	10.8	10.5	98.1	95.45
TNZ	80	6	4.8	10.8	10.7	10.6	99.07	98.1
	100	6	6	12	12.2	12.1	100	100.8
	120	6	7.2	13.2	13.0	13.1	98.4	99.2

Table 3. Analysis of tablet formulation for method A & Method

Method	Label Claim		Amount Found		% Recovery	
	CPH	TNZ	CPH	TNZ	CPH	TNZ
A	500	600	496	598	99.2 %	99.6 %
B	500	600	490	605	98 %	100.8 %

Table 4. Optical characteristics Data

Parameters	Method A		Method B	
	CPH	TNZ	CPH	TNZ
Analyticalwavelength	270nm and 320nm		270nm and 288nm	
Beer's range	0.2 –1.2µg/ml	10–60µg/ml	0.2 –1.2µg/ml	10-60µg/ml
Slope	0.52	0.01	0.51	0.009
Intercept	0.06	0.023	0.05	0.06
Correlationcoefficient	0.9995	0.997	0.999	0.998
Intra-dayprecision	0.366	0.951	0.350	0.421
Inter-day precision	0.433	0.853	0.465	0.243
LOD	0.310	0.180	0.101	0.212
LOQ	0.718	0.821	0.103	0.991
Tablet assay	99.2%	99.6%	98%	100.8%

Anaysis of the Tablet Formulation

Onconcurrent evaluation of CPH and TNZ in retail tablet dosage form (CIPLOX-TZ) containing retailproduct of CPH-500 mg and TNZ-600mg was completed. In this assay methodology, 20 tablets were squashed and ground to a fine powder. Powder comparable to 25 mg of CPH and 30 mg of TNZ was moved to a 50 ml volumetric flask containing around 50 ml of urea solution broke up and sonicated for 10 min. The solution was sifted through Whatmann filter paper. The solution was additionally dilution with D.W to get conc. of 10µg/ml for ciprofloxacin and 12µg/ml for tinidazole. Absorbance of these solution were estimated at 270nm(ciprofloxacin) and 320nm(tinidazole) as ϵ_1 and ϵ_2 and centralization of these two medications in the tablet were determined. The consequences of investigation were in table 3.

Validation Of Developed Method ^[14]

The developed U.V spectrophotometric was validated as per (International council for harmonization) guidelines in terms of linearity, range, specificity, precision, Accuracy, LOD & LOQ for estimation of ciprofloxacin and tinidazole in bulk & tablet dosage form.

*Linearity*¹⁵: From working standard of CPH(50µg/ml) 0.2, 0.4, 0.6, 0.8, 1.0, 1.2 and TNZ(50µg/ml) 10, 20, 30, 40, 50, 60 ml transferred to 10ml volumetric flask and volume was made up to mark with distilled water. These solution were then scanned in the range of 400- 200nm against solvent as blank at their respective wavelength and then calibration curve was plotted.

*Precision*¹⁶ Repeatability (inter day) and (intraday). In precision for interday we are required to analyses three samples per day for consecutive three days and for intraday one sample was analysed for three days.

*Accuracy*¹⁷: Known amount of standard of CPH and TNZ were added to the pre-analyzed sample solutions of tablet dosage forms. Results are shown in table 2.

LOD and LOQ^{18,19}: LOD and LOQ can be calculated as: LOD = 3.36/S
LOQ = 106/S

RESULT AND DISCUSSION

Solvency of ciprofloxine and Tinidazole

was viewed as in excess of 15 folds in 2.0M urea solution when compared with dissolvability in D.W. It is apparent from table 3 that the percent drug assessed in tablet definition was 99.2% for CPH and 99.6% for TNZ by technique An and 98% for CPH and 100.8% for TNZ by method IIThe optical characteristic such as, t , regression equation,beer's range and Correlation coefficient also determined in table 4.

CONCLUSION

The analytical study of Ciprofloxacin and Tinidazole by U.V Visible Spectrophotometry was successfully performed. Developed method enhances solubility of both water insoluble drugs and here was no involvementgenerated by urea in the estimation, therefore we analyzed that two UV spectroscopytechniques were established to be easy, accurate, inexpensive and speedy for concurrent estimation of CPH and TNZ in bulk and tablet dosage forms. So different dosage form containing CPH and TNZ can be easily analyzed by utilizing these techniques of spectroscopy. Ciprofloxine belongs to Class IV and Tinidazole belongs to Class II of BCS Classification.

Conflict of Interest

There is no conflict of interest

REFERENCES

1. Nidhi K, Indrajeet S, Khushboo M, et al, 2011. Hydrotropy: A promising tool for solubility enhancement: A review. *Int. J. Drug Dev.* 3(2), Pages- 26-33.
2. Sherje AP, Desai KJ, 2011. Spectrophotometric determination of poorly water soluble drug rosiglitazone using hydrotropic solubilization technique. *Indian journal of pharmaceutical sciences.*73(5), Pages- 579.
3. Poochikian.G, 1979. A reviewson the hydrptropes. *J Pharm Sci.*68, Pages- 728-729.
4. Maheshwari RK, 1979. Mixedhydrotropy in spectrophotometric analysis of poorly water-soluble drug. *Indian Pharmacist. Sci.*68, Pages- 728-9
5. Jain P, Goel A, Sharma S, Parmar M, 2010. Solubility enhancement techniques with special emphasis on hydrotrophy. *International Journal of Pharma Professional's Research.* Pages- 34-45
6. Shalu Verma, MohiniRawat, 2019. Novel Approach on Solubility Enhancement: Solid

- Dispersion a Brief Overview. *International Journal of Pharmacy and Biological Sciences.* 9(3), Pages- 1354-3272.
7. Mishra GP, Joshi HM, 2010. Quantitative estimation of Ciprofloxacin in marketed formulation by hydrotropic techniques. *International Journal of Pharmacy & Life Sciences*, Pages- 76.
 8. Maheshwari RK, Shukla RS, 2008. Quantitative spectrophotometric estimation of famotidine using hydrotropic solubilization technique. *Asian Journal of Chemistry.* 20(6), Page- 4221.
 9. Maheshwari RK, 2006. Solid dispersion and syrup formulation of poorly watersoluble drug by hydrotropy. *The Indian Pharmacist.* 5(50), Pages- 87-90.
 10. Bhatia MS, Kaskhedikar SG, Chaturvedi SC.1999; 61(5):311 High performance liquid chromatographic estimation of ciprofloxacin hydrochloride and tinidazole from tablets. *Indian Journal of Pharmaceutical Sciences.*
 11. Panzade PD, Mahadik KR, 2001. Simultaneous estimation of ofloxacin and tinidazole in tablet dosage form. *Indian Drug Bombay.* 38(7), Pages- 368-70.
 12. Tatane S, 2011. Development of UV spectrophotometric method of telmisartan in tablet formulation. *Journal of Advances in Pharmacy and Healthcare Research.* 1, Pages- 23-6.
 13. Rathod SD, Patil PM, Waghmare SS et al, 2012. UV-spectrophotometric method for estimation of telmisartan in bulk and tablet dosage form. *International Journal of Pharmaceutical Sciences and Research.* 3(10), Pages- 3936
 14. Kamberi M, Tsutsumi K, Kotegawa T et al, 1998. Determination of ciprofloxacin in plasma and urine by HPLC with ultraviolet detection. *Clinical chemistry.* 1; 44(6), Pages- 1251-5.
 15. Wu SS, Chein CY, Wen YH, 2008. Analysis of ciprofloxacin by a simple high- performance liquid chromatography method. *Journal of chromatographic science.* 46(6), Pages- 490-5.
 16. Krishna JR, Naga B, Sandhya SH et al, 2014. Development and Validation of RP-HPLC method for the Simultaneous estimation of Ciprofloxacin Hydrochloride and Ornidazole in Combined Pharmaceutical Dosage Form. *Journal of Advanced Pharmacy Education & Research.* 4(4), Page- 44.
 17. Grewal AS, Patro SK, Kanungo SK et al, 2012. Simultaneous Spectrophotometric estimation of Ciprofloxacin and Ornidazole in tablet dosage form. *International Journal of Pharmaceutical Sciences and Research.* 3(8), Pages- 2716.
 18. Ravisankar P, Navya CN, Pravallika D et al, 2015. A review on step-by-step analytical method validation. *IOSR J Pharm.* 5(10), Pages- 7-19.
 19. Virani P, Jain P, Raj H et al, 2014. Updated review: validation and method validation parameters. *PharmaTutor.*2(10), Pages- 27-37.