Wound Healing Activity of Cotton Fabrics Loaded with Silver Nanoparticles in Experimental Model of Diabetes

Jihan Hussein^{1*}, Mona El Bana¹, Yasmin Abdel Latif¹, Safaa Saleh² and Emad Tolba³

¹Medical Biochemistry Department, National Research Centre, Doki, Giza, Egypt. ²Branch of Biophysics, Faculty of Science (Girls), Al Azhar University, Egypt. ³Polymers and Pigments Department, National Research Centre, 33 El-Bohouth st., Dokki, Giza, Egypt. *Corresponding Author E-mail: jihan_husein@yahoo.co

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Diabetes is connected with diminished wound healing, that makes patients liable to continuing difficult wounds. Metal nanomaterials as single conjugates have established to keep possible properties of wound when metal nanoparticles are coupled with other wound covering materials. This study aimed to investigate a possible role of cotton fabrics full with silver nanoparticles (AgNPs) to enhance wound healing in diabetic model induced by streptozotocin (STZ). Animals were classified into four groups including the wounded group that were equivalently covered with sterile dressing that made of cotton fabric which had been saturated with different concentrations of silver nanoparticles, and the control group that was preserved with only cotton covering without any treatment (blank group); percent of wound contraction in different studied groups was estimated. Plasma nitric oxide (NO), malodialdehyde (MDA), reduced glutathione (GSH) were measured. Serum neutrophil elastase and nuclear factor kappa b (NF-?b) were assayed by ELISA. Homocystiene (Hcy) was estimated by HPLC. Our results revealed an elevation in wound area, MDA, NF-?b, Hcy, and elastase in the wound group compared to treated groups concomitant with a decrease in plasma nitric oxide and reduced glutathione activities, while treatment with AgNPs significantly ameliorated these parameters in treated group compared to blank group. AgNPs showed high wound contraction rates according to their used concentration .In conclusion; AgNPs have gained considerable attention amongst researchers in wound healing applications, owing to their physicochemical and biological properties. AgNPs promote wound healing and effectively control the growth of microorganisms at the wound site, and this strategy plays an important role in the treatment of wounds.

Keywords: Diabetes Mellitus; HPLC; Homocystiene; Silver Nanoparticles; Wound Healing.

Wound healing is a forced response to various types of stimuli that affects skin or any organ. In the case of tissue or skin injury, a sequential overlapping cascade of events occurs which eventually were resulted in the restoration of normal tissue¹. Wound healing is typically divided into different phases, namely inflammation, cell proliferation, angiogenesis, collagen deposition, and re-epithelization ^{2,3}. Impaired wound healing can lead to difficulties in treating deep tissue infections ¹. There are many human diseases characterized by impaired wound healing and chronic skin ulcers⁴. Thus, chronic wounds are a common complication in patients with diabetes

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that often lead to resection. Diabetes is associated with impaired wound healing, making patients susceptible to chronic non-healing wounds ⁵. Such wounds preceded 84% of all diabetic lower extremity amputations⁵⁻⁶ and once amputation occurs, patients have a 5-year mortality rate of 50%^{5, 7}.

Chronic diabetic wounds are trapped in a persistent inflammatory state with elevated levels of pro-inflammatory cytokines and proteases together with impaired expression of growth factors ⁵, in addition to elevation of oxidative stress state when the production of ROS exceeds the anti-oxidant capacity in diabetic patients 8. The formation of advanced glycation end-products (AGEs) under hyperglycemia and the interaction with their receptors (RAGE) are associated with impaired wound healing in diabetic mice as well⁹ . Neutrophils are the predominant cell type in the first inflammation phase and begin to shrink after 24-36 h by apoptosis in the time of circulating monocytes enter the wound and mature into tissue macrophages that play a very important role in the wound healing. This process is mediated by the chemokines IL-8 which is released by neutrophils, then attracts the macrophages and other cells to the wound site ¹⁰. Additionally, the enhancement of tumor necrosis factor -á (TNF-á) & IL-6 production induced nerve destruction¹¹

Indeed, skin wounds are typically treated with the combination of protective barrier, antibiotics and topical growth agents to shorten inflammatory phase which is a well-known symptom for chronic wounds. Although various wound healing agents are widely used, it has recently been shown that the topical drug delivery systems using nanoantibiotics (nAbts) represents a new paradigm involving nanomaterials to fight against microbial infection compared to the traditional antibiotic agents ¹². Thus, preparation of metal-based antimicrobial agents has drawn a significant attention in the past few years, as alternative and reports indicated that microbial resistance to the practiced antibiotics can be reversed by active nanometals13 . Metal-based nanoparticles including silver (AgNPs), gold (AuNPs), zinc oxide (ZnONPs), titiunm oxide (TiO₂NP) and copper oxide (CuONPs) have been shown to exhibit remarkable strong and sustainable antimicrobial activity against a wide array of bacteria, fungi, algae and even virus ¹⁴ and ¹⁵. The metal-based nanoparticles involve distinct multiple bactericidal mechanisms including oxidative stress, destroying bacterial cell membrane or other cellular organelles and non-specific mechanism (i.e. disruption of cell signaling process), which (match the color) evidently underline the powerful potential of nanosize metal-based particles as effective alternative antibacterial agents to avoid bacteria antibiotic-resistance mechanism^{16, 17}.

Among various metal nanomaterials, silver nanoparticles (AgNPs), which are widely utilized in formulating ointments for hurts ¹⁸,¹⁹. AgNPs are well known for their comprehensive range of antimicrobial activity, competently abolishing bacteria, viruses , fungi, protozoa and ²⁰. In in vitro study, AgNP was used for treatment of human keratinocytes and fibroblasts, and it effectively reduced inflammatory cytokines, oxidative stress and stimulated healing²¹ Additionally, the current application of AgNP reduced the counts of neutrophils and also interleukin (IL)-6 concentration, and linked with the elevation of vascular endothelial growth factor (VEGF), IL-10, and TGF-â levels ²². From these findings, we aimed to investigate a possible role of cotton fabrics loaded with silver nanoparticles in improving wound healing in experimental model of diabetes.

MATERIALS AND METHODS

Materials

Hcy (HPLC standard) was obtained from Sigma-Aldrich Co. (St. Louis, MO, USA), low molecular weight CMC (89% degree of deacetylation), and silver nitrate (AgNO₃) were bought from Sigma-Aldrich Co. (St. Louis, MO, USA). Traditional wound coverings were obtained from a local provider. Analytical grade anhydrous ethanol was purchased from Elnaser Company Egypt. All the reagents were used as received. **Animals**

Forty male albino rats weighing 180±10 g were used. All rats were housed separately in sterile stainless steel cages for one week in standard conditions of temperature and light; fed standard rodent chow. Animal procedures were performed in accordance with the Ethics Committee of the National Research Centre and followed the

recommendations of the National Institutes of Health Guide for Care and Use of Laboratory Animals (Publication No. 85-23, revised 1985). **Methods**

Synthesis of silver colloidal solutions using chemical reduction

Carboxymethyl Chitosan (CMC) containing Silver NPs was prepared in waterethanol mixture. In brief, one gram of CMC powder was dissolved in 50 ml solution with continuous string for 1h. Then, about 10/ mL of freshly prepared AgNO3 (0.2/ M) was added drop wise into CMC aqueous solution. The color changes to dark brown after adding silver nitrate solution. The mixture solution was kept at room temperature for 30/ min in dark. Then, 10 ml of anhydrous ethanol were slowly added to the CMC/AgNO3 mixture. The mixture was vigorously stirred and heated to reach 80°C (1°C/min) holding for 3 h until obtaining a homogenous colloidal solution without aggregating. The color of the solution was gradually changed from dark brown to yellowish brown indicating the formation of Ag NPs.

Cotton fabric treatment with CMC/silver colloidal solutions

In order to obtain the desired wound dressings, the traditional wound dressings were introduced in the CMC-silver nanoparticles colloidal solution, until the nanoparticles were incorporated in the wound dressing. Then, the wound dressings were removed from the solution and genially wiped over two filter paper to remove the excess of CMC/AgNPs solution. Finally, the wound dressings were soaked in ethanol solution for 3h and then collected on a glass Petri dish to dry at 60°C.

Characterization of as-prepared CMC/Ag NPs sample

Synthesized AgNPs absorption peak was observed in the UV-visible spectrophotometer (Shimadzu (UV 2500), Japan). UV-visible spectrophotometer range was from 200/ nm to 900/ nm. The TEM technique was employed to visualize the morphology and the size of the synthesized Ag NPs using Ultra High Resolution Transmission Electron Microscope(TEM), JEOL-2010). TEM grids were prepared by placing a drop of the particle solution on a carbon-coated copper grid and drying under lamp. In addition, the wound dressing treated samples were coated with gold and examined using field emission Scanning Electron Microscope SEM (Jeol JXA 840, Japan).

Study of antibacterial potential

The antibacterial activity of wound dressing samples was performed in nutrient agar medium against P. aeruginosa (Gram-negative bacteria: ATCC 433) and S. aureus: (Grampositive bacteria: ATCC 1688), the bacteria most commonly isolated from wound infections. In details, the medium was prepared and sterilized at 120 °C for 1 h in an autoclave and then it was incubated at 50 °C (0.5 mL/50 mL of medium) with the suspension (52.2 cfu mL"1) of microorganism (McFarland barium sulfate standard) and transferred into petri dish . Both bacteria were added separately into broth solution in separate test tubes. The dry sample was added to the dishes; the resulting growth of bacteria was observed when the plates were pre-incubated for 24 h at 37 °C for antibacterial activities. The existence of inhibition zones was measured as the distance from the border of disk to the edge of the bacterial growth.

Biological studies

In vitro study

Cell viability was assessed by the mitochondrial dependent reduction of yellow MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) to purple formazan (Mosmann,1983).

Procedure

All the following procedures were done in a sterile area using a Laminar flow biosafety cabinet Class II A2 (Manufactured by: Labconco).Cells were suspended in DMEM medium, 1% antibioticantimycotic mixture (10,000U/ml Potassium Penicillin, 10,000 μ g/ml Streptomycin Sulfate and 25 μ g/ml Amphotericin B) and 1% L-glutamine and 5% fetal bovine serum at 37 °C under 5% CO2 using CO2 incubator (Sartorius stedium ,biotech).

Human skin fibroblast (HSF) cell lines were purchased from American Type Culture Collection (ATCC) and batch cultured for 10 days, then seeded at a concentration of 10x103 cells/well in fresh complete growth medium in 96-well plastic plates at 37 °C for 24 h under 5% CO2 either alone (negative control) or with different concentrations of silver nanoparticles (500, 250, 125, 62.5, 31.25, 15.625 ug/ml). After 48 h of incubation, medium was aspirated, 20 ul MTT salt (2.5ig/ml) were added to each well and incubated for further four hours at 37°C under 5% CO2. To stop the reaction and dissolving the formed crystals, 200iL of 10% sodium dodecyl sulphate (SDS) in 0.01M HCL was added to each well and incubated overnight at $37^{\circ}C^{23}$.

The absorbance was then measured using a microplate multi-well reader (Bio-Rad Laboratories Inc., model 3350, Hercules, California, USA) at 595nm and a reference wavelength of 620nm.

Viability = absorbance of drug / absorbance of control x 100 Cutatoxiaity = 100, viability

Cytotoxicity = 100- viability

In vivo study Induction of diab

Induction of diabetes

Streptozotocin was dissolved in 50 mM sodium citrate solution (pH 4.5) containing 150 mM sodium chloride; the solution (6 mg/100 g body weight) was subcutaneously administered in rats. Fasting blood sugar was estimated after 3 days to confirm the development of diabetes mellitus. The animals were considered diabetic if fasting glucose level was \geq 200mg/dl ²⁴.

Incision wound operation

Diabetic rats were anesthetized with diethyl ether (40%) and their dorsal surface was shaved with a sterile blade. The shaved area was sterilized with ethanol (70%). A single longitudinal skin incision was done on the shaved area.

Experimental design

The wounded skin animals were uniformly dressed with experimental sterile dressing made of cotton fabric which had been impregnated with different concentrations of AgNPs. The negative control group was maintained with only cotton dressing without any treatment (Blank group). This experiment was conducted for a period of 7 days. Every day, the animals were dressed with a new cotton fabric impregnated with the grouprespective concentration of AgNPs. Fasting blood sugar was recorded on the initial and final day of the study.

The diabetic wounded skin animals were uniformly dressed with experimental sterile dressing made of cotton fabric which had been impregnated with different concentrations of AgNPs as follow:

Group I: Diabetic wounded rats treated with only



Fig. 1. Characterization of the fabricated AgNPs colloidal solutions: (A) UV–visible spectrophotometer range was from 250/ nm to 700/ nm, (B) Color change of CMCs/AgNO₃ solution after the addition of Ethanol, (C) TEM image of CMCs/Ag NP solution. Scale bar represents 100 nm.

cotton dressing (Blank group)

Group II: Diabetic wounded rats treated with the CMC-silver nanoparticles colloidal at concentration 130.676 ig /ml?

Group III: Diabetic wounded rats treated with Ag NP-based dressing at with the CMC-silver nanoparticles colloidal at concentration $65.338 \mu g /ml$.

Group IV: Diabetic wounded rats treated with cotton dressing *treatment with CMC/silver colloidal solutions* at concentration $32.669 \,\mu$ g/ml. Measurement of the wound area

The progressive changes in wound area (mm) were recorded on the 1st, 4th and 7th days along the experimental period. The size of the wound area was photographed and is presented



Fig. 2. Represents SEM micrographs of wound dressing materials before and after treatments. a and b) show smooth fiber structure without any contaminating particles on their surfaces. c, d ,e, and f show the wound dressing treated with CMCs/Ag NPs was covered with ultrafine silver nanoparticles .

in Fig1. The percentage of wound contraction was also calculated from the equation that was described previously 25 .

Wound contraction % = [Wound area day 0 wound area day n / Wound area day 0] ×100

n : number of days

After the experimental period, blood was withdrawn from the retro-orbital venous plexus of the eye using heparinized capillary tubes and collected in two tubes; the first one contains sodium fluoride for determination of fasting blood sugar

Table 1. Cytotoxicity of different concentration of Ag

Different dilution	Viability	Cytotoxicity	IC ₅₀	
of drug 1	%	%	(µg/	
ml)				
500 ug/ml	80.34	19.66	1306.76	
250 ug/ml	100	0		
125ug/ml	100	0		
62.5 ug/ml	100	0		
31.25 ug/ml	100	0		

IC₅₀: Lethal concentration of the sample which causes the death of 50% of cells in 48 hours

and the second tube was dry clean tube without anti-coagulant for serum separation. All tubes were centrifuged at 5000 rpm using cooling centrifuge (Laborzentrifugen, 2K15, Sigma, Germany) for 10 min. Serum was separated and stored at "20°C until determination of biochemical parameters.

Biochemical assays

Blood glucose was measured according to the method described by Trinder ²⁶, nitric oxide level (NO) was measured as nitrite and determined using Griess reagent, according to the method of Moshage et al., 27 . Lipid peroxidation was assayed by measuring the level of malondialdehyde using the previous method ²⁸ in which the thiobarbituric acid reactive substances (TBARS) react with thiobarbituric acid to produce a red colored complex having peak absorbance at 532 nm. Serum neutrophil elastase and NF-êb were measured by ELISA according to the method described by the manufacturing kit.

Determination of serum Homocystiene (hcy)

Hcy estimated by high-performance liquid chromatography (HPLC) system, Agilent technologies 1100 series equipped with a quaternary pump (G131A model) according to the method described previously ²⁹.



Fig.3. Agar diffusion test of wound dressing treated samples S. aureus (Gram-positive bacteria) and P. aeruginosa (Gram-negative bacteria)

Sample extraction: deproteinization of all samples was achieved using trichloro acetic acid (2% TCA).

HPLC condition: 30 il from each extracted samples were injected into HPLC; separation was accomplished on reversed phase column (C18 x 25 cm x 0.46 cm, and I.D. 5 im). The mobile phase was consisted of 40 mmol/L sodium phosphate monobasic monohydrate, 8 mmol/L heptanesulfonic acid, and 18% (v/v) methanol; pH of the mobile phase was adjusted to 3.1 then filtered 2 times through a 0.45 im membrane filter (0.45 µ m pore size, WCN type). The mobile phase was then delivered at a flow rate of 1 ml/min at 40°C. UV detection performed at 260 nm. Serial dilutions of hey standard were injected into HPLC. The concentrations in samples were calculated from the standard curve constructed by plotting peak areas versus the corresponding concentrations using Agilent Chem Station software for LC and LC/MC system (Agilent Technologies).

Statistical Analysis

All experiments were done in triplicate, and the results were presented as mean \pm standard deviation. Statistical analyses were performed with the one-way ANOVA test, by using Sigma Stat 3.5 software (Dundas Software Ltd, Toronto; Canada). P values ≥ 0.05 were considered statistically significant.

RESULTS AND DISCUSSION

Characterization of the CMCs-Ag NPs and wound dressing materials

Design and development of new composite materials including metal/polymer nanocomposites have attracted much attention in medicine due to their novel physicochemical features and improved biological performances in vivo, upon comparing with bulk forms. Numerous studies have been reported on the preparation of silver nanoparticles in the presence of polymeric materials, as stabilizing agents, to provide a high colloidal stability of AgNPs during and after the synthesis process (Kumar; et al. 2019). In this study, CMCs/Silver nanoparticles colloidal solutions were synthesized using anhydrous ethanol, as chemical reducing agent. The UV-Vis spectra of synthesized AgNPs are shown in Figure 1-A. The formation of AgNPs after the reaction with AgNO, solution was identified from the peak at around 430 nm due to the plasmon resonance formation. In addition, Figure 1-C represents the morphology and size of the prepared AgNPs confirmed by TEM analysis. The synthesis method produce nanoparticle sizes distributed at around 50-200 nm, it is important to mention that this methods tend to produce AgNPs with hexagonal-like forms.

The wound dressing materials were treated with CMCs-AgNPs colloidal solutions by a simple



Fig. 4. Viability versus different concentration of AgNPs

immersion method. The surface morphology of the wound dressing sample was investigated by SEM as shown in Fig. 2.Images of raw wound dressing (Fig. 2a and b) shows smooth longitudinalfiber structure without any contaminatingparticles on their surfaces. While, the wound dressing treated with CMCs/Ag NPs was covered with ultrafine silver nanoparticles (Fig. 2c and d). Antibacteril test

In the current study, the antimicrobial activity of wound dressing materials was assessed against two different pathogens: S. aureus (Gram-

Parameters Groups	Wound contraction (%) after one day	Wound contraction (%) after four days	Wound contraction (%) after seven days	
Group I(Blank group)	5	22	48	
Group II (treated group 1)	7	39ª	78.9ª	
Group III (treated group 2)	8	45ª	87.6 ^{ab}	
Group IV (treated group 3)	8	44 ^a	93.5 ^{ab}	

Table 2. Wounded skin contraction in experimental rats in days1,4 and 7

Pa value significant difference compared to group I. Pb value significant difference compared to group II. Pc value significant difference compared to group III.

	Day 1	Day4	Day7
Group I	6		
Group II			
Group III			
Group IV			

Fig. 5. Wound healing stages in the different studied groups in days1,4 and 7

Parameters Groups	Hcys mg/ml	Neutrophil elastase ng/ml	NF-κB ng/ml
Group I	13.2±0.7	15.4±0.9	5.7±0.1
Group II	10.8±1.5 ª	12.6±0.3 ª	4.5±0.2 ª
Group III	$6.0{\pm}0.5^{ab}$	11.7±0.3 ª	3.5±0.1 ^{ab}
Group IV	2.5 ± 0.4^{abc}	8.5±1.2 ^{abc}	$2.9{\pm}0.1^{abc}$

Table 3. Inflammatory markers in different studied groups

Pa value significant difference compared to group I.

Pb value significant difference compared to group II.

Pc value significant difference compared to group III.

Parameters Groups	NO μmol/g	MDA µmole/gm	GSH μmol/g	
Group I	9.7±0.2	181.5±7.6	38.7±2.2	
Group II	7.9±0.1ª	151.0±7.8 ª	53.5±1.7	
Group III	6.4 ± 0.6^{ab}	133.75±2.4 ª	70.2±1.1 ^{ab}	
Group IV	5.0±0.3 ^{abc}	120.2±4.9 ^{ab}	74.9±1.4 ^{ab}	

Pa value significant difference compared to group I.

Pb value significant difference compared to group II.

Pc value significant difference compared to group III.

positive bacteria) and P. aeruginosa (Gramnegative bacteria)

Cell viability

Cytotoxicity of the nanoparticles was carrying out and indicated that the highest concentration (500 ug/ml) is the only concentration appeared cytotoxicity (19.66%) while other concentrations appeared 0 toxicity; In addition, IC50 was calculated and it equal to 1306.76 ug/ ml (table 1, fig.4). Accordingly we used the high dose which is tenth the IC50 equal to 130.676 ig/ ml, and the other doses equal 65.338, and 32.669 ug/ml.

In vivo wound assay

In diabetic subjects, the wound healing process takes a long time and it is impaired rather than prevented ².A non-healing wound is prone to produce a lot of complications which considered an important factor in delaying healing process. These complications include functional limitations, including alteration in gait and difficulty in walking; infections like cellulitis, abscess, osteomyelitis; gangrene and septicaemia and possible malignant changes in some cases ³⁰. Chronic wounds are at risk of developing malignant changes, known as a Marjolin's ulcer, an aggressive form of squamous cell carcinoma³¹.

In the current study the percentages of wound contraction in diabetic group that was maintained with only cotton dressing without any treatment (blank group) was only 48% (table 2, fig 4).

Thus, diabetes is a disease of altered glucose homeostasis and persistent hyperglycemia lead to advanced glycation end product (AGE) which is primarily responsible for the damage of cells and have a slow turn over. This hyperglycemia leads to elevation of reactive oxygen species (ROS) production and releasing of cytochrome C followed by activation of caspase-3and myocardial cell apoptosis ^{32, 33}. Partial inhibition of increased glucose levels by insulin almost completely prevents myocardial cell death; otherwise, it could be argued that there is a significant increase in apoptosis with hyperglycemia. Deregulation of apoptosis in response to hyperglycemia is

generalized all over the body, leading to impaired wound healing along with the involvement of other target organs ³⁴.

Formation of AGEs under hyperglycemia elicited production of ROS by binding to its receptor (RAGE) expressed in various skin cells including keratinocytes, fibroblasts, dendrocytes, and to a lesser extent in endothelial cells and mononuclear cells⁸, and induce the activation of NF-êB ³⁵ which results in pathological gene expression and further impedes the normal activity of these cells during wound healing and impaired wound healing ³⁶.

Uncontrolled diabetes leads to wound infections and inûammation due to high levels of pro-inûammatory cytokines such as TNF-á, NF-êB and IL- 6 (Wang et al., 2012). The elevated levels of these cytokines were linked to delay healing causing lesser migration and proliferation of fibroblasts and keratinocytes, reduced accumulation of collagen and deferred re-epithelialization ³⁷.

In agreement, the current study appeared that, the inflammatory biomarkers homocysteine, neutrophil elastase, and NF-êB in group I represented inflammatory chronicity whereas a lesser degree of inflammation was seen in wounds treated with AgNPs in treated groups I, II, and III (table 3).

Nanotechnology is a multidisciplinary scientific field that has drawn worldwide attention from various researchers in science and industry. Nanotechnology offers the facile synthesis of metal-based biocompatible nanomaterials that can be applied to a wide range of potential applications in medical and biological sciences ³⁸. In wound healing, nanoparticles have a broad range of applications and they offer a novel solution in wound care³⁹

In the current study the percentages of wound contraction in treated groups (group II, III and IV) were significantly higher as recorded (78.9%, 87.6% and 93.5%) respectively; that is mean the high concentration of AgNPs gave the best effect and showed high wound contraction rates than the lower concentrations; thus the healing of these wound in our study appeared as a dose dependent (table 2, fig 5).

AgNPs offer a high-degree of biocompatibility and biodegradability in physiological conditions and can be considered as an effective material for wound dressings in the treatment of different wounds ³⁹. Thus, silver nanoparticles (AgNPs) are classified as metalbased nanoparticles and have gained considerable attention amongst researchers in wound healing applications, owing to their physicochemical and biological properties. AgNPs are non-cytotoxic and safe for patients in wound care management. The unique intrinsic features of AgNPs promote wound healing and effectively control the growth of microorganisms at the wound site, and this strategy plays an important role in the treatment of both acute and chronic wounds.

In the current study there was a significant increase in antioxidant enzyme activities represented by GSH with a significant decrease in MDA and NO levels in the treated groups compared with the blank group (table 4). In addition, these results were in agreement with the work of Arora et al., ⁴⁰ who investigated the effects of silver particles of 7-20 nm on primary mouse cells and observed an increase in levels of reduced GSH and superoxide dismutase (SOD). The authors hypothesized that, at lower concentrations of silver (up to 25ig/ml and 100ig/ml), there was an enhancement of cellular protection mechanisms; in their earlier work on secondary skin cell lines, this enhancement was restricted to lower doses of silver, but it is noteworthy that the non-toxic concentration range is comparable to that encountered in topical applications of silver products, providing evidence for the safety of silver nanomaterials at such doses 40

Inflammatory biomarkers presented as homocystien, neutrophil elastase and NF-êB in group I representing inflammatory chronicity whereas a lesser degree of inflammation was seen in wounds treated with AgNPs in treated groups (table 3). In addition there is significant differences between treated groups and each other indicating a recognizable effect of the high concentration of nanoparticles in group 4 in compared to other treated groups (group 2 and 3). Thus, AgNPs are believed to decrease the time for fibroblast's invasion into wound tissue, and also possess antiinflammatory properties ⁴¹ .AgNPs are known as a protease inactivator that acts to decrease inflammation and also reduce the time for tissue formation ⁴², which is considered an important factor in wound healing among diabetic subjects.

CONCLUSION

Actually, metal nanomaterials, as single conjugates have proven to possess potential wound healing properties; when metal nanoparticles are coupled with other wound dressing materials, it effectively removes microbes from the wound site. The output of this study has reported that treatment with cotton fabrics loaded with silver nanoparticles AgNPs resulted in a significant improvement in wound healing *in vivo* through increased antioxidant activity and decreased inflammatory markers in chronic disease causing hyper proliferative wounds.

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