# A flavonoid out of *Tephrosia purpurea* extract and its antimicrobial effect

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#### **ABSTRACT**

Flavonoids are present widely in Angiospermic plants and are associated with several pharmacological activities including antibacterial, antidiabetic, antifungal, anti-allergic and anti-oxidant activities. Flavonoids because of their low toxicity compared to other active plant compounds can be ingested in significant quantities by human beings in their diet. Flavonoid can initially prevent the ageing, cancer and inflammation effectively rather than cure because they inhibit the occurrence of oxidative stress in the cell. Tolerance to ultraviolet and altered behavior of T, B, NH and mast cells and neutraphills in their presence assure the activation of immune system. Body's action against them as foreign compound induces phase II enzyme which are mutagens and carcinogens eliminating. The present article shows the significance of flavonoids as potential drug, emphasizing their extraction out of Indian ethnomedicinal plant *Tephrosia purpurea* because of its easy availability and study of MIC value of particular flavonoid found effective on very much prevailed microbe born diseases.

Key words: Immunomodulatory, ethnomedicinal, minimum inhibitory concentration. Antimicrobial.

## INTRODUCTION

The production of free radicals (reactive oxygen containing molecule) in all in energy producing mechanism result in oxidative stress linked with cancer, ageing, atherosclerosis, inflammation and neurodegenerative diseases. Flavonoids are actively involved in the detoxification of highly reactive species.

Detoxification of hydrogen peroxide in vacuole has been proposed by Yamaski H in 1997.

2FlaOH + H2O2 = 2 Flavo\* + 2 H2O

# Flavonoid Peroxidase Flavonoid phenoxyl radical

The resulting flavonoid Phenoxyl radical can react with ascorbic acid (ASA) to generate the monodehydroascorbic acid radical (MDA\*).

2 Flavo\* + 2 ASA = 2 FlavOH + 2 MDA\*

# MDA\* can in turn nonenzymatically generate ASA and DHA (Dehydorascorbate reductase)

Tephrosia purpurea of family Leguminosae are very rich in flavonoid. It is a small annual shrub used in the treatment of asthma. Roots are useful to cure cuts and wounds. Whole plant is used in spleen enlargement and stem used as tooth brush cure many dental disorder. It cures colic pain when used with sendha salt.

Flavonoids are synthesized by the phenylpropanoid metabolic path way in which the amino acid phenylalanine is used to produce 4-coumaroyl-CoA.

### **MATERIAL AND METHODS**

### **Plant Material**

The plant material (near about 2 Kg)

collected from local surroundings after weighing, washed thoroughly and dried in air at room temperature for more than one month. The dried plant material was grinded to powder to about 40-60 mesh size. Then it was weighted and stored carefully in bottles for extraction and chemical testing.

# **Extraction and Isolation**

Extraction was carried out in Soxhlet apparatus in the laboratory with 90% ethanol as solvent. The extract was concentrated in vaccum evaporator to a good yield of 6.3%.

The biologically active compound was separated from the crude extract by column chromatography using EtoAc: ccl4 solvent system

as mobile phase and silica gel as stationary phase.

The compound was than isolated followed by TLC using same solvent combination for that fraction in column.

#### **RESULTS**

The tentative structure of compound was elucidated as Mf.  $C_{21}H_{20}O_{11}$ .

Molecular Formula -3, 7, 2' trihydroxy 6'ì – O –  $^2$  – D glucopyranosyl flavone Molecular Formula –  $C_{21}H_{21}O_{11}$  Molecular Weight – 449 M.P. – 210 – 212  $^{\circ}C$ 

It was light greenish amorphous powder

Table 1: Showing the weight of plant material after drying and percentage loss in weight of plant material of *Tephrosia purpuria* 

S. No.	Description	Weight in gms.
1.	Wt. Of plant material in wet fresh condition	240 gms.
2.	Wt. Of plant material after drying condition	160 gms.
3.	Loss in weight on drying	240-160 = 80 gms.
4.	Percentage loss after drying	40%

Table 2: Percentage yield of Tephrosia purpuria using different solvents in Soxhlet apparatus

S. No.	Solvent used	Wt. of Powedered Material	Volume of solvent	Wt. of extract	Character of extract	Percentage of yeild
1.	90% Ethanol	160 gms.	750 ml.	7.20 ml.		6.3%
2.	Water	160 gms.	500 ml.	8.00 ml.		7.5%

Table 3: TLC of 90% alcoholic extract of the plant Tephrosia purpuria

S. No.	Solvent System	Ratio	Visual light	lodine Chamber	Rf Value	Fraction Code
1.	EtoAc: CCI <sub>4</sub>	1:1	Blackish greer	Dark Green	0.92	FR - 1
2.	CHCI <sub>3</sub> : MeOH	2:1	Olive g	reen Dark Green	0.86	FR - 2

separated as fraction-1 from column. It was blackish green colored compound with Rf value 0.92 on TLC was followed by acid hydrolysis, methylation and glycosidal test.

# IR

3524 (OH), 1650 (C = O  $- \gamma$  - pyrone), 1590, 1565, 1515 (C = C aromatic)

#### **PMR**

(Py) -8.35 (3H, d, J = 9.6, H - 5, 2', 6'), 7.15 (1 H, br. s, 3 - OH), 7.44 (4H, m, aromatic protons) 10.30 (1H, br. s, 7 - OH), 9.80 (1H, br. s, 4' - OH), 13.62 (1H, s, 3 - OH).

On analysing under different analytical procedure showed maximum peak in ultraviolet spectrum at 270 mm.

# **DISCUSSION**

The known compound 3,7,2' – trihydroxy 6'- O -  $\beta$  – D glucopyranosyl flavone was identified by CoTLC with authentic sample and structure was

Table 5: Minimum inhibitory concentration (MIC) value of different fraction of flavonoids compound of Tephrosia purpurea against Staphylococcus aureus MTCC 96

S.No.	Fraction	MIC
1.	FR- 1	0.72
2.	FR-2	0.34

Table 4: Chemical Properties of Tephrosia purpurea

S. No	Properties	Percentage
1.	Ash content	30
2.	Water soluble ash content	70
3.	Acid soluble ash content	4.2
4.	Fiber value	Not more
		than 30
5.	Saponification value	68



established by comparison of measured UV, NMR and mass spectral data with spectroscopic data available from literature. 3,7,2' – Trihydroxy 2'- O - <sup>2</sup> – D glucopyranosyl flavone showed <sup>1</sup>H and 13CNMR data in full agreement with those given by Malikov and Yuldashev (2002).

The chemical structure of flavanonols includes 3- Hydroxyflavanones or Dihydroflavonols. They contain two asymmetric carbon atoms. Flavanols differed from flavanones by the presence of strong peaks in their mass spectra. It is demonstrated that flavanonols exhibit a base peak for [M-57] <sup>+</sup> ion whereas a peak for the [M-43] <sup>+</sup> ion was typical of flavones. It should be noted that an alcoholic solution of flavanonols at room temperature can undergo cis-trans epimerization to form a new isomeric compound.

Several such compounds have been isolated from plants of different families. Genistein and Daidzein soy flavonoid appear to reduce the risk of prostrate and breast cancer, Hesperidins raises the blood levels of the good high-density lipoprotein and lowers the bad low-density lipoprotein and triglycerides. It also possesses significant anti-inflammatory and analgesic effect.

Quercetin can inhibit the growth of head and neck cancers and can stop reverse transcriptase, the method HIV uses to replicate itself. It has demonstrated significant anti-inflammatory because of direct inhibition of several initial process of inflammation. As it inhibits both the production

and release of histamine and other allergic inflammatory mediators. Tangeretin induces apoptosis programmed cell death in leukemia cells but does not harm normal cells Prenylnaringenin are very potential antioxidant. Epicatechin improves blood flows and thus seems good for cardiac cells. The flavon compound is isolated from *Tephrosia Purpurea* was found to be anti-microbial when tested in experimental rats. A detail study of this compound is needed to be established.

#### **Biological activity**

The flavone isolated from alcoholic extract of stems and leaves of *Tephrosia purpurea* showed antimicrobial activity against Staphylococcus aureus. The bacterial strain used for the present study was procured from M.T.C.C. Chandigarh.

For antimicrobial activity disc diffusion was followed. The agar medium was used. The plates were incubated at 37 °C over night and examined for zones of growth inhibition.

The minimum inhibitory concentration of the extract was determined for Staphylococcus aureus using the two fold serial micro dilution method with saline at a final concentration ranging from 10,000 mg / ml. to 0.0024 mg / ml. The MIC values were taken as the lowest concentration of the extracts in the wells of the microtiter plate. These plates were interpreted as visible growth of microorganisms. Thus growth of bacteria is noticed inhibited with moderate zone of inhibition.

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