

Metalloantibiotics-III

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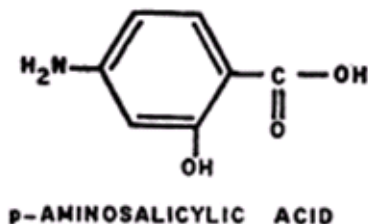
ABSTRACT

Metal ions after forming complexes with an antibiotics alters the antimicrobial activity of an antibiotics alone.

Key words: Metal ions, antibiotics

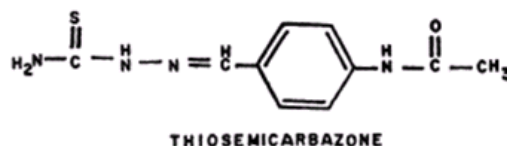
INTRODUCTION

The antitubercular activity of such drugs as INH, PAS, and thiosemicarbazone is affected by the ability of the compounds to form complexes with Cu²⁺. The Cu²⁺ complex of PAS was later found to be as active as PAS, but the effectiveness of both substances is enhanced by excess Cu²⁺ (89). The Cu²⁺ as well as the Fe²⁺, Co²⁺, Mn²⁺, and Cr²⁺ complexes of PAS are less tuberculostatic than the pure compound. Other investigators reported that Co²⁺ enhances (103), and that the majority of metallic ions tested suppress (17), the antitubercular activity of PAS. The toxicity of Cu²⁺ or Hg²⁺ for frog heart tissue is suppressed by PAS (28) but the Fe²⁺ complex is more toxic towards mice than is PAS itself (39). INH and PAS, thiosemicarbazone suppresses the toxicity of Cu²⁺ and Hg²⁺ towards isolated frog heart tissue (28).



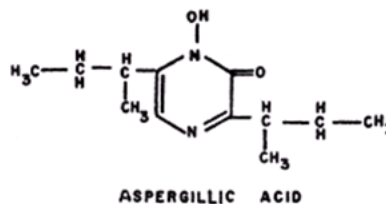
Aspergillilic acid

The antimicrobial activity of aspergillilic acid is suppressed by Fe⁺⁺ and enhanced by Bi³⁺. The potency of the compound is also enhanced by Co²⁺, Ni²⁺, Zn²⁺, and As³⁺, but to a lesser extent than by Bi³⁺ (46, 47). Goth (45) has postulated that aspergillilic acid is antitubercular because of its ability to chelate Fe⁺⁺ which he observed to be essential for growth of mycobacteria.



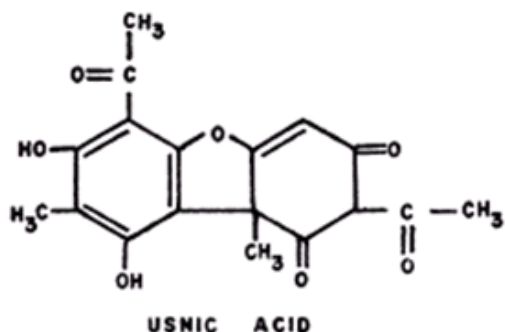
Usnic acid

Johnson et al. (57) have suggested that usnic acid may owe its antitubercular action to inhibition of phosphorylation reactions, because antimicrobial concentrations of the compound suppress oxidative phosphorylation of rat liver and kidney homogenates.

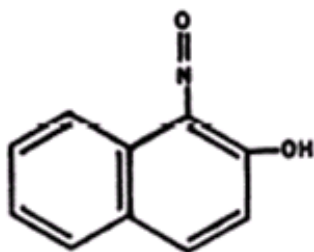


Miscellaneous antimicrobial compounds

Compounds whose Antimicrobial Activity is Influenced by Metallic Cations. The inhibitory effect of penicillin and bacitracin on the growth of *Micrococcus pyogenes* enhanced by Co^{2+} , (104); and the inhibition by 2,3-dimercaptopropanol



(British anti-Lewisite) of the binding of penicillin by whole cells is counteracted by Co^{2+} and Mn^{2+} (30). Bacitracin has been observed to be precipitated by salts of heavy metals and to be partially inactivated by British anti-Lewisite (11). Mg^{2+} , Mn^{++} , Ca^{++} , and Fe^{++} suppress the ability of polymyxin to inhibit *Pseudomonas aeruginosa*; a Mg^{2+} :drug ratio of 400:1 is required for maximum activity of Mg^{2+} (73). The ability of novobiocin to inhibit gram negative bacteria is suppressed by Mg^{2+} and, to a lesser extent by Ca^{2+} , Sr^{4+} , and Ba^{3+} (19, 117). Co^{2+} slightly enhances the ability of chloramphenicol to inhibit growth of *B. subtilis*. An analogue of chloramphenicol, 3,4-dichlorophenylserine, is inert against strains of *Enterobacteriaceae* but prevents sporulation of *Aspergillus niger* by the irreversible binding of essential Cu^{2+} (108). The ability of 1-

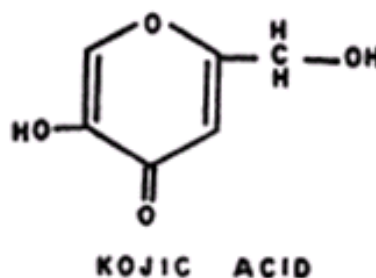


nitroso-2-naphthol to inhibit growth of gram positive bacteria is suppressed by Fe^{2+} and slightly enhanced by Cu^{2+} and Co^{2+} ; Ni^{2+} , Zn^{2+} , and Mn^{2+} are inactive.

Kojic acid

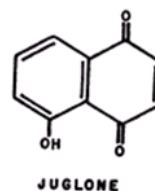
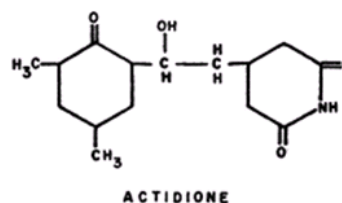
Inhibits *P. fluorescens* (but not *B. subtilis*) more strongly in the presence of certain concentrations of Al^{+++} , Cu^{++} , or Fe^{2+} ; similarly, Stove (100) noted that metal derivatives of kojic acid are better fungicides than is the nonchelated substance.

The ability of actidione to inhibit growth of *Saccharomyces cerevisiae* is suppressed by oxalate; and the growth inhibitory action of juglone towards *B. subtilis* is strongly enhanced by Hg^{2+} and Co^{2+} , moderately enhanced by Al^{+++} , Ni^{++} , Cu^{2+} , Zn^{2+} , and Mn^{2+} , and suppressed by citrate, tartrate, and oxalate.



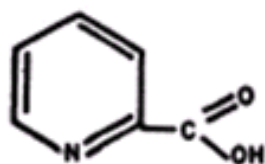
A-Picolinic acid

When tested with *B. subtilis*, is suppressed strongly by Co^{2+} or Ni^{2+} and moderately by Cu^{2+}

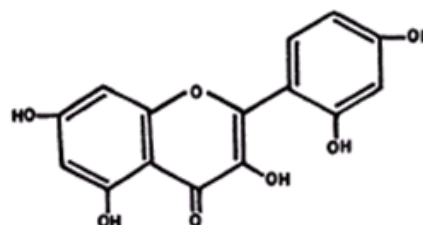


or Zn^{2+} . With *P. fluorescens*, the ions are inactive except for Cu^{++} , which possesses slight reversing ability. However, the toxicity of high concentrations of Co^{2+} towards cells of each species is suppressed

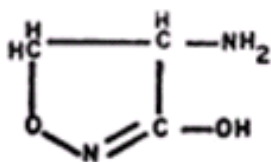
by α -picolinic acid (118). Previous studies demonstrated that Co^{2+} , but not Cu^{2+} , suppresses the antitubercular activity of α -picolinic acid hydrazide (34) and that Cu^{2+} , Fe^{2+} , Mn^{2+} , and



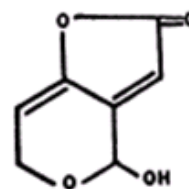
**α -PICOLINIC
ACID**



MORIN



CYCLOSERINE



PATULIN

Mg^{2+} suppress the inhibition of metallo-enzymes of rice plants by α -picolinic and fusarinic (5-butylicpicolinic) acids (102).

The ability of cycloserine (oxamycin) to inhibit growth of *B. subtilis* is suppressed by citrate, tartrate, oxalate, phosphate, or EDTA, whereas the antagonism of the substance towards growth of *P. fluorescens* is suppressed only by Cu^{2+} (118).

Antagonism of morin towards *B. subtilis* is, like that of cycloserine, suppressed by citrate, tartrate, or oxalate; the inhibition by morin of *P. fluorescens* is suppressed strongly by Co^{++} , and to a lesser extent by Fe^{++} , oxalate, or phosphate (118).

The activity of patulin is suppressed by Mn^{2+} , Ca^{2+} , and Ba^{3+} , and slightly enhanced by Sb^{+} , Co^{2+} , Cd^{2+} , Ni^{2+} , and Zn^{2+} .

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